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## A Direct Determination of the Molecular Structure of Prostaglandin $F_{2-1}$

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#### (Received 9 July 1962)

The tri-*p*-bromobenzoate of the methyl ester of prostaglandin  $F_{2-1}(C_{42}H_{47}O_8Br_3)$  is orthorhombic with  $a = 26 \cdot 14$ ,  $b = 33 \cdot 93$  and  $c = 4 \cdot 76$  Å. The space group is  $P2_12_12_1$  with Z = 4. The structure was solved by heavy-atom methods and refined by three-dimensional anisotropic least-squares treatment to give the molecular structure and configuration of prostaglandin  $F_{2-1}$ , which belongs to a new class of hormones having smooth-muscle-stimulating and blood-pressure-depressing activity.

#### Introduction

A smooth-muscle-stimulating and blood-pressuredepressing activity was discovered in human semen by Goldblatt (1933) and by von Euler (1934, 1935) who also found a similar activity in sheep sperm and named the factor 'prostaglandin'.

In 1960 Bergström & Sjövall (1960) isolated two active crystalline compounds  $PGE_1(C_{20}H_{34}O_5)$  and  $PGF_{1-1}(C_{20}H_{36}O_5)$  from sheep prostate glands. Reduction of PGE1 with borohydride (Bergström et al., 1962) yielded  $PGF_{1-1}$  together with the isomeric compound  $PGF_{2-1}$ .

The tri-p-bromo- and tri-p-iodobenzoates of the methyl ester of  $PGF_{2-1}$  were prepared by Prof. S. Bergström who kindly provided them for an X-ray analysis in order to determine ab initio the molecular structures.

### Experimental

The two benzoates crystallize as beautiful needles from ethanol. All crystals studied gave excellent X-ray photographs, which indicated that the two derivatives are isomorphous. The unit cell is orthorhombic with space g . The cell dimensions are:

group $P2_12_12_1$
--------------------

A C 16 - 27

	a	ь	c
Br-derivative	26·14 Å	33∙93 Å	4·76 Å
I-derivative	26·49	34∙39	4·76

Full three-dimensional data (Cu  $K\alpha$  radiation) were collected for the bromo compound. Multiple-film Weissenberg photographs were recorded for the 0-2 layers about c (needle direction) and the 0-5 layers about a. Only hk0 intensities were estimated for the iodo derivative. The intensities were corrected for the Lorentz and polarization factors but not for absorption.

## Structure determination

The initial work was done on the (001) projection. From a sharpened Patterson projection of the Iderivative, the relative positions of the heavy atoms of the four molecules per cell were obtained by simple vector addition and subtraction. The origin was then determined from the known symmetry of the projection. As expected the Patterson synthesis was completely dominated by I-I vectors (Fig. 1).

An electron-density projection showed most of the atoms of the molecule (Fig. 2) but also contained some spurious detail. This, however, was still present,

when more atoms, e.g. those of the benzoate groups were included in the analysis and the general appearance of the maps did not improve much from that of Fig. 2. Similar effects were encountered with the Br-derivative hk0 data. They were caused by the large anisotropic vibrations of the heavy atoms. As soon as these were accounted for a projection could be found which was chemically sensible and which did refine. Anisotropic least-squares treatment of the Br-derivative projection data reduced R to 20%.

At that stage the three-dimensional data (1776



Fig. 1. Patterson (001) projection for the I-derivative. Crosses mark heavy atom vectors calculated from the final bromine positions.

 $\begin{array}{c} O_1 \\ O_2 \\ O_3 \\ O_4 \\ O_5 \\ O_6 \\ O_7 \\ O_8 \\ C_2 \\ C_3 \\ C_4 \\ C_5 \\ C_6 \\ C_7 \\ C_8 \\ C_9 \\ C_{11} \\ C_{12} \\ C_{13} \\ C_{14} \\ C_{15} \\ C_{14} \\ C_{15} \\ C_{16} \\ C_{15} \\ C_{16} \\ C_{16}$ 



Fig. 2. Iodine-phased electron density projection along the c-axis with the correct structure outlined. Contours at arbitrary but equal levels except for the heavy atom, which has different contour intervals.

independent reflexions) were available. A Patterson synthesis sharpened to point atoms at rest (Abrahamsson & Maslen, 1963) was calculated. The resolution in the *c*-direction was still low due to the lack of high orders of *l*. This, and the presence of mirror planes, caused some trouble in finding the z-coordinates of the heavy atoms (see Fig. 3). Coordinates chosen to give the best compromise solution to the Patterson later proved to be surprisingly accurate (within 0·17 Å).

Table 1. Atomic coordinates

	x/a	y/b	z/c		x/a	u/b	zlc
Br <sub>1</sub>	0.63174	0.47212	0.01758	C	0.59564	0.30800	0.10090
$Br_2$	0.63476	0.35299	-0.21371	C17	0.60688	0.27266	- 0.24244
$Br_3$	0.29836	0.19886	0.16054		0.57848	0.23038	0.15469
01	0.34242	0.12472	-0.51624	C <sub>20</sub>	0.47724	0.03968	-0.10408
$O_2$	0.37917	0.08772	-0.17795	C <sub>at</sub>	0.49046	0.00306	0.08600
$O_3$	0.51648	0.17763	-0.03904		0.49111	0.47009	0.99090
0 <b>4</b>	0.47548	0.21145	0.27676		0.48345	0.43917	
0 <sub>5</sub>	0.39422	0.33494	-0.06161		0.45817	0.20094	- 0.40800
O <sub>6</sub>	0.42681	0.29515	-0.37597	C	0.45393	0.36086	- 0.22800
07	0.64002	0.16036	0.40039		0.42221	0.22957	
O <sub>8</sub>	0.68815	0.12388	0.11804		0.39942	0.96075	-0.23383
$C_1$	0.19402	0.05198	-0.11009		0.55798	0.00400	-0.20072
$C_2$	0.19771	0.08252	-0.28724		0.60572	0.00706	0.01940
C <sub>3</sub>	0.24290	0.09588	-0.37575		0.65203	0.09038	0.15000
C <sub>4</sub>	0.28700	0.08330	-0.24648	C <sub>10</sub>	0.67528	0.15901	0.10999
$C_5$	0.28642	0.05488	-0.04880		0.70620	0.10000	0.24200
C <sub>6</sub>	0.23859	0.03885	0.02603		0.60709	0.99707	0.13500
Č,	0.33866	0.10228	-0.32869	Č <sup>33</sup>	0.79200	0.22707	0.24057
C.	0.42725	0.10450	-0.23528	C <sup>34</sup>	0.76164	0.20200	0.14650
C.	0.43282	0.14464	-0.10760	C35	0.77960	0.20030	-0.03000
Cin	0.48675	0.14279	0.04589	C <sup>36</sup>	0.74697	0.21976	-0.14132
C,	0.51282	0.10674	-0.10328	C <sub>37</sub>	0.69959	0.18892	-0.03900
C <sub>1</sub>	0.46710	0.07703	-0.12347	C <sup>38</sup>	0.65506	0.02368	0.07943
C,	0.50569	0.20957	0.10708	C <sup>39</sup>	0.01010	0.01565	0.14440
C1	0.54020	0.24402	0.02812	C40	0.31319	0.47755	-0.56601
C.,	0.52861	0.28079	0.12427	C41	0.33840	0.44075	-0.64926
C	0.55678	0.31405	0.10016	C42	0.30974	0.40358	-0.57723
- 10	0 00010	0.01100	0.10010				

#### SIXTEN ABRAHAMSSON

#### Table 2. Vibrational parameters

Allowance was made for anisotropic vibrations with  $2^{-(h^{2b}_{11}+k^{2b}_{22}+l^{2b}_{33}+klb_{23}+hlb_{13}+hkb_{12})}$ 

			·			
	$b_{11}$	$b_{22}$	$b_{33}$	$b_{23}$	$b_{13}$	$b_{12}$
Br.	0.00293	0.00392	0.29367	-0.00325	-0.01480	0.00287
$Br_{1}$	0.00878	0.00236	0.28466	0.00437	0.00991	- 0.00608
Br.	0.00431	0.00167	0.24959	0.00221	0.00609	0.00184
0.	0.00262	0.00154	0.16485	0.00902	0.00414	0.00097
$\stackrel{\circ}{0}$	0.00241	0.00158	0.04023	0.00450	-0.00858	- 0.00018
$\tilde{O}_{2}^{2}$	0.00208	0.00095	0.10525	-0.00587	0.00649	- 0.00086
0 <u>3</u>	0.00521	0.00114	0.19606	-0.00059	0.00333	- 0.00070
õ <b>4</b>	0.00409	0.00170	0.15280	0.00683	0.00976	-0.00251
0.	0.00555	0.00149	0.17847	0.00366	0.00426	0.00074
0.	0.00406	0.00239	0.23268	-0.00167	0.01554	- 0.00369
ŏ7	0.00220	0.00206	0.03925	0.00169	0.00254	- 0.00108
$\tilde{C}^{8}$	0.00474	0.00172	0.18647	- 0.00406	0.02437	- 0.00206
$C^1$	0.00161	0.00347	0.23723	- 0.00858	0.00886	- 0.00250
$C^2$	0.00390	0.00918	0.33806	0.00360	0.00505	-0.00178
$C^3$	0.00345	0.00130	0.26335	- 0.01230	- 0.00462	0.00110
$C_4$	0.00240	0.00164	0.26255	-0.00572	0.01158	0.00199
$C_{5}$	0.00254	0.00317	0.20200	0.01304	0.00974	0.00100
$\tilde{C}_{6}$	0.00107	0.00134	0.17379	0.01941	0.00274	-0.00199
C,	0.00197	0.00102	0.08023	0.00271	-0.01110	0.000137
$C^8$	0.00120	0.00103	0.14233	0.00371	0.00034	-0.00115
C <sup>9</sup>	0.00185	0.00093	0.12216	-0.00430	0.00569	- 0.00113
C <sup>10</sup>	0.00130	0.00074	0.20169	-0.00698	0.00799	- 0.00108
$C^{11}$	0.00139	0.00074	0.17588	0.01210	0.00733	- 0.00072
$C^{12}$	0.00190	0.00106	0.21504	0.00526	-0.00160	0.00104
C13	0.00210	0.00100	0.07704	0.00545	0.00767	0.00104
C14	0.00310	0.00135	0.20704		0.00200	- 0.00304
C <sup>15</sup>	0.00427	0.00115	0.06204	0.000247	0.00390	- 0.00095
C16	0.00202	0.00160	0.10506	- 0.00083	- 0.00200	- 0.00248
017	0.00539	0.00152	0.99604	0.01874	0.01575	- 0.00129
C18	0.00514	0.00000	0.23034	- 0.00166	0.01070	- 0.00250
C <sub>19</sub>	0.00914	0.00081	0.122429	-0.00100	0.01270	0.00079
C <sub>20</sub>	0.00207	0.00084	0.94077	0.00140		- 0.00038
C <sup>21</sup>	0.00334	0.00090	0.11509	-0.00104	-0.02731	0.00151
$C_{22}$	0.00400	0.00106	0.19940	-0.00293	- 0.00019	0.00049
C <sub>23</sub>	0.00256	0.00110	0.12249	-0.01200	-0.00228	- 0.00114
C <sub>24</sub>	0.00330	0.000119	0.96669	-0.01377	- 0.01040	- 0.00047
C <sub>25</sub>	0.00341	0.00109	0.00002	0.00114	0.00778	- 0.00088
C <sub>26</sub>	0.00270	0.00002	0.00223	-0.000112	0.01002	0.00041
C <sub>27</sub>	0.00390	0.00090	0.15666	0.00033	0.00248	- 0.00201
C <sub>28</sub>	0.00240	0.00022	0.10002	0.00467	- 0.00348	- 0.00004
C <sub>29</sub>	0.00212	0.00092	0.15615	0.00407	. 0.00600	- 0.00008
C <sub>30</sub>	0.00240	0.00001	0.20024	-0.00493	-0.00088	-0.00004
C <sub>31</sub>	0.00232	0.00140	0.14792	0.00940	-0.02243	-0.00128
C32	0.00331	0.00140	0.14703	0.00400	0.01044	-0.00053
C <sub>33</sub>	0.00487	0.00106	0.26037	-0.01966	0.02831	-0.00182
C34	0.00963	0.00020	0.20741	0.00464	0.00284	- 0.00085
035	0.00996	0.00146	0.20971	0.00600	0.01000	-0.00082
$C_{36}$	0.00979	0.00140	0.20281	-0.00009	0.09749	- 0.00405
C37	0.00245	0.00100	0.10454	-0.00107	0.02743	0.00014
C38	0.00400	0.000120	0.10007	0.00699	- 0.09190	-0.00076
C <sub>39</sub>	0.00961	0.00033	0.10404	- 0.00038	0.02130	0.00032
040	0.00201	0.00121	0.19404	- 0.00080	- 0.02013	0.00009
041	0.00469	0.00129	0.21233	- 0.0094 F	0.00702	0.00145
U42	0.00408	0.00113	0.20210	0.00345	-0.02783	-0.00103

The three-dimensional electron-density series calculated on the bromine phases showed the complete molecule as found from the (001) projection. Two rounds of Fourier refinement, during which leastsquares methods were applied to the heavy atoms, reduced R to 23% for all observed reflexions and confirmed the assignment of atomic numbers to the peaks in the projection analysis.

All atoms were finally included in the anisotropic least-squares refinement. Three cycles reduced R to 10.9%. It was not considered worth while to perform further 7-hour least-squares cycles as the molecular

structure is fully established with the accuracy in positions reached. However, the shifts in the least accurately placed atoms (the benzene-ring carbons) approached the standard deviations.

For carbon and oxygen the scattering curves of Berghuis *et al.* (1955) were used and for bromine that of Thomas & Umeda (1957) corrected for the f' component of the anomalous dispersion (Dauben & Templeton, 1955). The calculations were performed on a Ferranti-Mercury computer with programs described by Mills & Rollett (1961). The weighting scheme adopted in the least-squares refinement was

Table 3. Interatomic distances and bond angles

	Bond	l distances	
Bond		Bond	
$Br_1-C_1$	1·88 Å	C22-C24	1·44 Å
$Br_{9}-C_{17}$	1.88	C C.	1.33
Bra-Cas	1.90	$C_{0\pi}^{*} - C_{0\pi}^{*}$	1.39
C <sub>1</sub> –C <sub>9</sub>	1.34	$C_{22}^{33} - C_{27}^{30}$	1.36
$C_{2} - C_{3}$	1.33	$C_{27} - C_{22}$	1.33
$C_{3} - C_{4}$	1.37	C <sub>o</sub> -C <sub>o</sub>	1.50
$C_4 - C_5$	1.35	Co-Cio	1.59
$C_5 - C_6$	1.41	$C_{10} - \hat{C}_{11}$	1.57
$C_1 - C_6$	1.40	$C_{11} - C_{12}$	1.57
$C_{4} - C_{7}$	1.55	$C_{12}^{}-C_{20}^{}$	1.50
$C_{7} - O_{1}$	1.18	$C_{20} - C_{21}$	1.58
$C_{7} - O_{2}$	1.37	$C_{21} - C_{22}$	1.52
$O_2 - C_8$	1.41	$C_{22} - C_{23}$	1.53
$C_{10} - O_3$	1.47	$C_{23} - C_{24}$	1.56
$O_{3} - C_{13}$	1.32	$C_{24} - C_{25}$	1.59
$C_{13} - O_4$	1.13	$C_{25} - C_{26}$	1.54
$C_{13} - C_{14}$	1.52	$C_{26}^{-}-O_{5}^{-}$	1.10
$C_{14} - C_{15}$	1.36	$C_{26} - O_{6}$	l·44
$C_{15} - C_{16}$	1.35	O <sub>6</sub> -C <sub>27</sub>	1.49
C <sub>16</sub> –C <sub>17</sub>	1.43	$C_{11} - C_{28}$	1.58
C <sub>17</sub> C <sub>18</sub>	1.42	$C_{28} - C_{29}$	1.36
C <sub>18</sub> -C <sub>19</sub>	1.42	$C_{29} - C_{30}$	1.48
$C_{14} - C_{19}$	1.34	C <sub>30</sub> -C <sub>38</sub>	1.54
C30-O8	1.49	$C_{38} - C_{39}$	1.52
$O_{8} - C_{31}$	1.34	$C_{39} - C_{40}$	1.58
$O_{7} - C_{31}$	1.19	$C_{40} - C_{41}$	1.47
$C_{31} - C_{32}$	1.45	$C_{41} - C_{42}$	1.51

		Bond ang	les		
Bond ang	le	Bond ang	le	Bond ang	gle
Br <sub>1</sub> -C <sub>1</sub> -C <sub>2</sub>	123°	$C_{18} - C_{17} - Br_{9}$	117°	C10-C11-C28	105°
$Br_1 - C_1 - C_6$	118	$C_{17}^{10} - C_{18}^{11} - C_{19}^{10}$	117	$C_{19}^{10} - C_{11}^{11} - C_{98}^{20}$	116
$C_{e}-C_{1}-C_{0}$	118	$C_{19} - C_{19} - C_{14}$	120	$C_{11} - C_{12} - C_{8}$	99
C,-C,-C	122	COC.	116	$C_{11} - C_{19} - C_{90}$	116
$C_{9} - C_{9} - C_{4}$	120	$O_{0}^{*}-C_{1}^{*}-O_{7}^{*}$	122	$C_{8} - C_{19} - C_{90}$	118
$C_{a} - C_{a} - C_{z}$	122	0C.,-C.,	110	C19-C90-C91	112
Co-Co-Co	119	0,-C,-C,-C,-	127	$C_{99}^{12} - C_{91}^{20} - C_{99}^{21}$	108
C <sub>0</sub> -C <sub>4</sub> -C <sub>5</sub>	119	C C C.	118	CCC.	112
CCC	118	Cal-Ca-Car	130	CCC.	112
$C_{r} - C_{r} - C_{r}$	120	$C_{00} - C_{00} - C_{00}$	112	$C_{00} - C_{04} - C_{05}$	108
O,-C,-C	122	C.,-C.,-C.,	125	$C_{04} - C_{05} - C_{06}$	103
$O_{0}-C_{7}-C_{4}$	113	CCC.	114	$C_{95} - C_{96} - O_{5}$	137
0,-C-0	124	$C_{a} - C_{a} - C_{a}$	124	$C_{05} - C_{06} - O_{6}$	104
$C_{n} - O_{n} - C_{n}$	116	$C_{a} - C_{a} - Br_{a}$	118	$O_{z} - C_{z} - O_{z}$	118
C, -O, -C,	114	$C_{ac} - C_{ac} - Br_{a}$	119	$C_{ne} - O_e - C_{ne}$	117
$O_{0} - C_{10} - O_{1}$	125	Car-Car-Car	116	$C_{11} - C_{00} - C_{00}$	116
$O_{0} - C_{10} - C_{10}$	112	$C_{a} - C_{a} - C_{a}$	128	$C_{00} - C_{00} - C_{00}$	121
$O_{1}-C_{1}-C_{1}$	123	$C_{10} - C_{0} - C_{0}$	111	$C_{00}^{20} - C_{00}^{20} - C_{00}^{30}$	116
$C_{10} - C_{14} - C_{15}$	119	$O_0 - C_0 - C_0$	112	$C_{00}^{29} - C_{00}^{-0} - O_{0}^{-0}$	108
$C_{10} - C_{14} - C_{10}$	120	O <sub>0</sub> -C <sub>0</sub> -C <sub>10</sub>	108	0CC.	104
$C_{10} - C_{14} - C_{15}$	121	$C_{0}^{2} - C_{0}^{0} - C_{10}^{12}$	104	$C_{n} - C_{n} - C_{n}$	112
$C_{14} - C_{15} - C_{16}$	128	CC1C1	102	Coo-Coo-Cao	113
$C_{14} - C_{10} - C_{10}$	110	$C_{0} - C_{10} - O_{0}$	108	Coo-Coo-Co	113
$C_{10} - C_{10} - C_{10}$	124	$C_{11} - C_{10} - O_{0}$	106	$C_{40} - C_{41} - C_{49}$	115
$C_{1,2}$ - $C_{1,2}$ - $Br_{2,2}$	118	$C_{1} - C_{1} - C_{1}$	101	40 41 42	
~16 ~17 D <sup>1</sup> 2	110	$\sim_{10}$ $\sim_{11}$ $\sim_{12}$	T V T		

$$w = \frac{1}{1 + [(|F_o| - 8|F_{\min}|)/5|F_{\min}|]^2}.$$

### Discussion

The coordinates of the atoms in the molecule other than hydrogen are given in Table 1, the vibrational parameters in Table 2. The final structure factors are listed in Table 6. A composite drawing of the three-



Fig. 3. Sections of the three-dimensional point atoms at rest. Patterson synthesis for the bromo derivative. (a) Harker section at z=1/2, (b) section at z=26/80. Crosses mark Br-Br vectors calculated from the final positions.

dimensional electron-density series based on the final phases is shown in Fig. 4.

Table 4.	Average	values	for	different	classes
0	f bond di	istances	an	d angles	

	Bond		Angle
Benzene rings	$\mathbf{C}-\mathbf{C}$	$1{\cdot}38\pm0{\cdot}05~{\rm \AA}$	$120 \pm 10^{\circ}$
Unstrained carbon single bonds	C-C	$1.54 \pm 0.07$	111 <u>+</u> 8
Carboxyl carbon to oxygen	C = 0 C - 0	$1.15 \pm 0.05$ $1.37 \pm 0.07$	
Other carbon-oxygen	C - O	$1.46 \pm 0.05$	
	Br-C	$1.89 \pm 0.01$	

Table 5	5. Some	short e	contacts	between	one	molecule
and	its equi	ivalent	repeate	d along	the c	-axis

$0_{1} - 0_{2}$	3·52 Å	$C_{15} - C_{18}$	3∙64 Å
$0_{3} - 0_{4}$	3.62	$C_{15}^{10} - C_{19}^{10}$	3.93
$0_{7} - 0_{8}$	3.84	$C_{16}^{10} - C_{18}^{10}$	3.67
$O_{5} - O_{6}$	3.63	$C_{33} - C_{36}$	3.55
$C_2 - C_6$	3.74	$C_{33} - C_{37}$	3.88
$C_{3} - C_{5}$	3.67	$C_{34} - C_{36}$	3.92
$C_3 - C_6$	3.44	$C_{8} - C_{10}$	3.98

The molecular structure of the tri-*p*-bromobenzoate derivative of the methyl ester of  $PGF_{2-1}(C_{42}H_{47}O_8Br_3)$  is all in accordance with the formula (I) for the free acid of  $PGF_{2-1}$  independently arrived at by Bergström *et al.* (1962*a*) using chemical and physicochemical methods other than X-ray diffraction.



 $PGF_{2-1}$  is thus a trihydroxy acid with two hydroxyl groups attached directly to a five membered hydrocarbon ring and the third along a carbon side chain

C32

C33

1.37

# SIXTEN ABRAHAMSSON

# Table 6. Final structure factors for the bromo derivative

An asterisk marks changes in h and k. Each non-asterisk line contains l,  $100A_o$ ,  $100B_o$ ,  $100A_c$  and  $100B_c$ 

PLANES

	a 1		1 12 1									
	3 0 - 3388 a	- 11 12	0 -400 -3584 -433	-14410	, , , , , , , , , , , , , , , , , , ,	، فرەر	1;017	· · ·	10 1	:	2.0	
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•	3 0 ~1693 0 ^ ~111 0 ~11 <b>6</b> 0	-751	1 3700 62370 2738 2 2920 -1273 2427	-1141	3 -1674 -233 3 -2929 1323	-1911	1 478		4 370 1 39	12	-4414	-744
	1 27261 0 30224 1 3220' 0 4235	•	0 -1415 1 -1107 4189 -1300	-2 47 5 49 3 2	-10;1 1 -368y 7189	- 3640	-;;;	· • •	0267 -854	1405	-744	1.270
•	, -1591 0 -1591 , 5 , 6 9864 0	• •	1 2012 1850 1910	-3136	3 3949 967	+ 255	-1111 641	• •	3364 17 666	-4090	a;t7	-4434
•		3376 .	· · · · · · · · · · · · · · · · · · ·		1 1333 718	707	3743 380	1	111	3 4 4 0 • • • • •	4485	-1100
	-4901 0 -4380 3836 0 3534	e 0	1 -3908 -4539 -4156	-4245	3 1185 3099 3 1409 -1579	1004	-1123	· · ·	3517	-160,	3703 -244	
•	5 1495 0 1356 0 7 1 0 11081 0	• •	1 -1775 1814 -1675 1 -1775 1814 -1675	1247	1 1585 6737	1 426	-7 496		3	2513	419	3090
•		1018	• • • • • • • • •	-3001 .	3 441 -1680	373	-1421	• •	1445	1186	3079	, <b>,, ,</b>
• •	-7826 0 -7928 1345 0 1062		° ° −1918 0 1 3633 −3363 4041	-3199	0 0 3515 1 -1565 1507 2 -1596 1446	-1363	3871 1506		1010 2373 21	352	1901-	3738
•	1 3596 0 1771 0 9 1 0 4019 0	• •	1 -113 -1010 -8, 1 -35 0 -113 -	-741 .	3 -3 0 43 - 637	-1430	-394	; :	463	1501	-857	1733
. 1	0 1158 0 0 2700 0	ijo6 >343 •	1 1078 3937 3317 7 36	3347	1 -+C1574 2 -4C55330	-1146	-1138	;	997	-192	** 13 3 3	
	4374 0 4564 -5981 0 -5267	•	1 318 1449 338 1 -1301 571 -1733	3351 ·	3 -1949 395 3 7 6 15463	-1781	259 23587	• •	***	- 358	1594	
. 3	1686 0 100	• •	• · · · · · · · · · · · · · · · · · · ·	1990	1 -11738 4744 3 -466 -384 3 979 1486	-11468 -652 815	4643	•••	100	::,	3517	
	0 6154 0 0 1535 0 0 -1936 0	3544 1350 •	1 1546 3971 1673 3 39 0 0 1454 D	3315 ·	3 - 560 3 - 71 35	-53 40	-6673	•	33	146 <u>5</u> .86	-1354	1357
•	-4033 0 -3954	• •	: -3341 čs -3389 : 30	- 66 •	1 3576 -8317	1468	-8508	•	300	1841	- 353	
	• 13 • 7323 •	6730 .	3 -1464 -338 -1837 1 31	-447 .	3473 -449 3 10 -7268	3845	- 302		-40	-1320	-443	-164
.;	0 1875 0	1715	1 1726 -Coj 1668	-1361 -776	1 -1736 6871 2 4117 -387	-3803 416%	70;2	• .• .	329	-1134	352	-1117
į	3968 0 4199 -5834 0 -5743 2111 0 2116	•	1 471 -992 743 2 894 526 793	400	4377 5126	4367	- ¥5 39 60 30		659 963	735	3964	
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•	o 17 0 3460 0	3300	675 0 323	• •	3 3350 -1157	1954	-1010		611 35	90	1950	\$7
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	5431 0 5693 0 19	•	1 1371 11699 1319 -4340 -3557 -4991	•	3 1183 1536 3 15 9 0 3411	763	991	• ; • '	12	• -6 107	-1514	-65-5
.;	0 ~5690 0 0 1597 0 0 30	-5861	2 1648 1372 1999 2 3 3 - 4012 0 - 1141	1165	1517 -1413	3498	-1 345		230	-1573	9319	6253 -1645
i	11691 0 10935 -1504 0 -1703		5474 4468 5346 -7956 5788 -1858	4193 ·	3 16 391	-747	957	:	-,;	,;;;, ,,,,,,	-,1	-23456
•	0 31 0 3536 0		0 -4001	• •	1 -0005 5030 3 -3135 -703 3 17	-3306	4807	• • • •	33	334	-1110	392 5380
:	-3807 0 -3395 -4450 0 -4157		-5930 5305 -9246 -3783 -3936 -3468 - 5	3493	° ° -5402 1 233 -1437 2 -2105 240	-131	-5397		359	;019 -4;83	4247	-41+3
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•	o 38 3148 o 3943 ⊃ 30	• • ;	4378 7347 3657 -1613 1685 -1515	6461 1776 •	1 2354 -2598 3 25 9 2012	3139	-2447	••••	°	-6246 2301		-5894
• ;	1945 0 3139 0 31 0 1416 0	0 0 1 1463 J	-35705 0 -24345 3453 -266 2265 2567 -1730 3041 -	-1]2 +	-305 -3507 3 34	-542	-1611	• ; • `	°, °	- 15 1 1	1070	-3513
· ·	0 23 0 23 0 24 0 0 24 0 0 25 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0	• •	-1996 o -1832		-1057 1360 3 45	-792	1019	<u></u>	736 929 -	3008	4509 -1890	3519
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	° 37 ° 967 •	793	4456 0 4743 3713 -1743 3314 -	3005	3 30 0 7757 1 3393 194	0 1979	60C0 473	• ; • '		1155 3516	-1823	40 48
_ ·	919 0 400 1 0	• .	-995 1611 -356	903 1097 •	-3 #7 -3 6 40049	-1823	9656	• • •	973 - 930	-547	-3873 843	-3718
• ;	1638 0 3340 0 -3948 0 1 I	-3345	4683 0 4403 3751 -7293 3328 - 3763 3547 3548	64y7 •	-1555 -718 3 -18 -148	-1603	-750	1 6		6.15	5054	- 3280
.;	349 -313 \$29 1548 1254 1538	-741 1135 • 3	1761 133 1437 3 13 -6105 0 -1326	10.	-397 3460	-413	1368	•••		3365		1714
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°.	0 -1844 0 4944 10915 5905	-1436 2 13032 1	1715 - 1756 - 5304 - 5889 - 5395 -	5984	3 30 3 30 3 30		-4363	1 7	596 -	7150	6931 1094	-0973
•••	-3490 7388 -3381 1603 1098 3139 4	7471 *	3033 -173 3334 3 13 -7047 0 -7581	-119 .	5 35 1168 1689 -601 450	1101	1593	• • •	15 163	1909 1240	- 1411	1920
;	-2391 0 -3351 -9630 -1233 -988 2025 -1190	-1606 1 -9145 1 1963 *	-751 -5975 2,03 -	1143 <b>•</b>	3 38 3979	•	3737	• , •	6	553	-1030	394
• ' •	-3694 778 -2557	558	-4885 0 -4373 1815 - 3003 2050 -	3 393	-4051 0 1859 0	-4017	•	1.1	36		-1357	-469
	7330 7330 6993 3837 3543 3763	7002	4633 -3309 4559 - 558 1873 598	32 42 0 1 36 2 1	1581 6198 7669	676	7755 .	· · ·		6916	878	6447 -911
î	-15443 -104 -15609	8460 ·	1 18 4870 0 4443	· • • • •	-3344 -1380 4 - 3 -11765 0 -	- 40 J	-1554	• • • *		1876	۵۵۱٤۵ ه	-931
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;	6895 -8704 6638 1536 2106 1432 -1691 1346 -1274	-8368 • 1964 • 1914 ·	3 19 1757 0 1605 -1311 3316 -1141	470	- 3546 - 11103	14310 -3473 -1	10873		- 6	4390	3176	- 3900
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1	-4398 10913 -4013 -3618 -3340 -3367 -11 -1758 -17	10186 1 -3084 •	1733 -3481 1999 -; 3 37 -2802 -3120	865	14447 5556	1 4 3 3 3	••••	· , ·	-		131	947
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÷.,	-1953 2645 -1489	-64C .	-,,,, -,,, 1033 ·	•	-104 -2245	-149 -	355	1.3	4	1114 112 -	-544	140
÷	-5633 -7354 -5326	-1362 -	1060 74 844	·	-5690 -596	-5325	-6.5	, -s	, ,, -	112	,	-1;;;
· .	1716 -513 697	-315 •	, ja 1355 1195 1405 1	11 <sup>8</sup> · ·	4505 -1039 3374 1770 4 13	350,	1041			002	- **3	,,,
	-5353 -3694 -5137 578 1381 581	-35,50 - 1 130,5 - 1	-081 y07 -843 1 3 36 386 -679 340 -1	140	-771 0 0403 -5015 -4037 -2031 -	-300 0860 -	· \$1,2	<u>_</u>	, , , ,	1 0 5 0 5 4 0 4	-001	-1003
3	1430 1727 1167	1651						1 6		- 0.0		

MOTECULAR	STRUCTURE	OF	PROSTAGLANDIN	F
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Table 6 (cont.)

	1869 -855	-41	-410	• 7 • 4	-2 40 9 0	-::::	• # 33 • 1403 304 2140 322 • 0 13	• 10 10			
1 -030	556	-1 326	859	+ 7 15	1838 -18	1817	• • • • • • • • • • • • • • • • • • •	- 1638 -	-1508 -	- 100	1339
a 20135	-4575	1775	- 41 18	1 -605 3 1378	1391 -363 1093 I354	557	• 9 · · · · · · · · · · · · · · · · · ·	0 1174 1 3160 -	1519	- 100	1356
• • • • • •		16198	-1001	• 7 16	70 48 0	7583	3 -937 0 +936 0 * 9 J	• 10 1t	-18:6 -		-10.48
· · · · · · · · · · · · · · · · · · ·	3194	-5056	11 11	• 7 17	-1107 1706	-1.4.	1 1008 -2069 1138 -2337 2 -490 4146 -479 4451	* 10 19	-398	3343	-390
1 -5968	-8100	-5771	-7930	1 -5988	97 -6675 -86 -1100	108 -73	• • • • • • • • • • • • • • • • • • •	1 1651	-568	1476	-5
• • • • • • • • • • • • • • • • • • • •		,84,	•,•	• 7 18	-3519	-1004	3 -2618 602 -2567 570 - 2 3	• 10 30	•	t 49	•
1 -14301	-880	-14776	-1903	• 7 1,	930 2016	10]]	6 6 7568 6 6716 1 5323 -1925 5215 -1886	1 1359	183	1143	193
a -3487 1 3731	-6416	-4143 3661	-6 3 3 1	1 -1797 3 4111	86 -1880	90 166	* 7 4 * 7 4 * 0 7541 0 6706	1 -1384	-576	-:489 2535	
• 6 S	-1900	-637	-3761	• 7 30	-42.42 0	-4077	: -7414 -1508 -7686 -1563 3 5644 2913 5463 1850	• • • • • • • • • • • • • • • • • • • •		-1061	
1 6014	-444	3510	-773 1195	1 -3144 2 -1000 - 7 31	520 -1030 1473 -3330	3704 374	2 1475 7115 1551 7484 1 150094 100763	• 10 24	-683	+ 395	-711
• • • • •	<b>^</b>	9453		1 450	-3640 0	-1481 -3388	• • • • • • • • • • • • • • • • • • •	1 176 3	-751	1961	
	3786	-555	1013	• 7 • •	1303 -1481	1193 2538	1 17; -1907 109 -1011 1 -1256 1555 -1755 1470	• :• :5	-,•,	2114	••••
1 -1815	1055	-18 41	1678	1 -1743	-1893 -1133	-1045	0 0 13976 0 1360, 1 854 1015 1405 1435	1 -1314	1650	-1463	1477
- 6 8	-930	-1931	-767	· 7 · 13	3675 -3582	4095	3 3397 -1375 3403 +1315 • 7 8 • 0 • 1477 • 1860	• 10 10 0 1360 1 -841	-697	1386	-415
· 6 32 46	-1955	\$795	-2625	1 - 108	-8918 0 -1859 -111	-80 Jo - Jo 49	1 -7334 (451 -5033 1698 3 -3314 356 -2942 316	• • • • • • • • • • • • • • • • • • • •	•	5 - 5 -	
1 1359	***7	831	-255	• 7 •5	5705 0 -1031 1666	5853	• 9 7 • • 3673 • 3341 1 -3111 -2126 -1510 -7485	• 10 10 • -1 478		-1331	••/•
• • <del>• • • • • • • • • • • • • • • • • </del>		-197	•	* 7 36	-1480 019	-1396	2 2693 3652 2850 3979 • 7 10	• 10 ±173	3937	3377	3068
- 6 11 - 1 100	-1053	-3005	-3570	÷ 997	4228 2041	-1694	0 0 11263 0 11203 1 -4145 -3244 -4463 -3493 1 -1665 6443 -3493	· · · · · · · · · · · · · · · · · · ·	-1034	-355	-978
1 1455	5228 -3586	378	4553 -3130	t -= 375 + -1394	-388 -3015 390 -1699	-119	* g 12 o o =1411o o =1198a	· 10 31	-437	-906	-438
0 -6115 1 2040	5403	-5288 #343	6 417	- 7 38	-935 -757	-811	1 -5004 5137 -5001 7300 1 394 1540 351 1373 0 13	6 -1466 1 3618 3 741	1845	2035	1861
· 6 13	3117	303	3401	1 - 35 3	1398 -479 -918 755	-968	0 0 -7415 0 -6421 1 4695 3337 4320 3061	* 10 31 0 -3214	. •	-1613	•
0 ~5989 1 750 2 8973	-1143	867 31 16	-3478	• 7 30	-3817 0	-13 37	3 -213 1314 -703 1049 • 3 13 • 0 0 7482 0 6030	• 10 33	-894	-1908	-915
• 6 14 • 851a		7178		1 1093	-64 1313	-17	1 525 2633 1011 3803 2 -1929 -3740 -316, -4214	* 10 34 3 1389	117	1913	337
-757	-1756	-5710	-1789	• 7 33	492 -1328	078	• • • • • • • • • • • • • • • • • • • •	10 15	- 2 5 5	-1473	-303
e 16194 1 1864	1623	14113 5948	3675	• 7 , 6	1001 05	1133	3 1470 -CC1 1014 -689 9 15	3 -1.1 37	-2750	-473	-3376
• 6 16	1178	799	1779	• 7 36	-1415 0	-1006	0 0 -2634 0 -8336 1 -3198 -933 -3534 -1639	• • • •	6343	-0708	6403
1 - 3787	1453	- 5991	1331	• • • 7710	0 7314		o o −3474 o −3935 • 9 17	• • • •	-3 340	\$37	-1846
• • • 17 • • • • • • • •		47 48			11879 0	1991	0 0 -5333 0 -4943 1 -1143 4639 -1374 4793	3114	-1656	-2855	-1519
• • •	-3137	952	-1111	• -5391 1 -5391	0 -10103 1971 -3339	1913	• 5 18 1 -1413 -729 -1342 -673	• • • •	915		t.,
1 3560	6103	3410	5943	• • • •	-4901 50	-4363	3 -3097 -746 -3373 -844 • 6 19 •	1 -1 376	3713	1458	3937
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1 1818	466	7073 7077	533	0 8546 1 3864	4300 1711	40.91	• 2 20 • 0 -2452 0 -2328 1 666 -2555 586 -2242	??	-3969	693	-3543
1 -1343	-978 -3148	-1116	-887 -2078	• • • • • •	0 4505	./•	2 866 495 659 377 • 0 21	1 -506.	6414	-1815	7099
• 6 11		3881		1 -3456 2 3344	737 -3039	693	0 0 -1950 0 -1756 3 1548 -1317 2325 -1043		-13 30	-3467	-1193
- 6 33	969	-1345	943	• • • • • • • • • • • • • • • • • • •	0 6085 1897 -1405	0 1474	0 0	• .: 7	-3327	• • •	-10860
• -5987 1 -985	1663	-5507		- 3 -4133	-83 -4006	-79	2 -1880 -2000 -1733 -1901 • 4 23	-1944	-153	-;190 79	1572
· 6 25	1 300	- 17 5 1	1039	0 9359 1 0175	1634 6368	1658	0 0 1071 0 000 1 200 −3088 973 −3073 3 −1360 −412 −1050 −343	1 3762	- 5897	3690	-8050 -17
1 318	2589	2603	1386	• t 7 0 -10372	0 -10881	•	* y 24 2 2040 -yt8 2326 -2039	• 11 9	-1324	1807	-1080
	141	-4876		1 -2288 2 -1337	5003 -3355 3884 -1178	4931	• 6 33 0 0 -1230 0 -1196 1 -110 1891 -272 1416	· -1165	4205	-3370	4409
• 6 16 1 - 535	-1460	-411	-1790	0 9299 1 -7985	0 8273 -183 -7328	-165	1 -1066 -146 -1133 -160 4 16	• 11 10	-13831		-12946
• 6 • 1181 • 6 • 17	643 0	-1040	\$65	• • • • • •	196 -3344	312	o o −1250 o −684 t 1575 −1153 1584 −1159 3 100 1610 441 1788		-1693	-3189	-1000
· 6	-1318	-347	-1485	1 5493	-1481 -997	-1396	5 9 27 C 0 1634 0 1596	1459	-2674	3477	-3307
-1106	-374	-1093	-165	• -2876 1 -218	311 -1345	435	1 -1481 1046 -1174 839 2 C5 -1510 70 -1336	·	-1070	-968	-1133
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# SIXTEN ABRAHAMSSON

Table 6 (cont.)

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Table 6 (cont.)

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Fig. 4. Composite drawing of the final electron density series. Contours given at intervals of 1 e.Å<sup>-3</sup> starting with 2 e.Å<sup>-3</sup> except for the bromine atoms which are contoured at intervals of 5 e.Å<sup>-3</sup>.

containing a *trans* double bond. The carboxyl group is at the end of a second side chain. The chemistry will be fully discussed elsewhere by Prof. S. Bergström and co-workers.

The X-ray analysis, in addition, gives the stereochemistry of the molecule which would probably have been difficult to determine with conventional chemical methods. Oxidative ozonolysis of  $PGF_{2-1}$  yields  $\alpha$ -hydroxyheptylic acid with  $(\alpha)_D^{23} + 9 \cdot 0^\circ$  in sodium hydroxide. According to Baker & Meister (1951) the acid has then the 2D configuration. The absolute configuration of the molecule is thereby known and is given correctly in Fig. 5.

 $PGF_{1-1}$  which is formed at the same time as  $PGF_{2-1}$  on reduction of  $PGE_1$ , which has a carbonyl oxygen instead of the hydroxyl group adjacent to the carboxyl side chain (I) (Bergström *et al.*, 1962*a*), differs from  $PGF_{2-1}$  only in the steric position of this hydroxyl group.

The bond distances and angles in the molecule are given in Table 3. The standard deviations were calculated from the diagonal elements of the inverse normal-equations matrix (Ahmed & Cruickshank, 1953; Darlow, 1960). The average values are for bromine:  $\sigma(x) = \sigma(y) = 0.005$  Å,  $\sigma(z) = 0.008$  Å, for oxygen:  $\sigma(x) = \sigma(y) = 0.015$  Å,  $\sigma(z) = 0.025$  Å, and for those carbon atoms which are least accurate:  $\sigma(x) =$  $\sigma(y) = 0.02$  Å,  $\sigma(z) = 0.05$  Å. The accuracy appears



Fig. 5. Spacial diagram of the molecule (correct absolute configuration) with the numbering of atoms indicated.

slightly better if one compares distances and angles with the corresponding generally accepted values but is, of course, still low in a heavily vibrating structure of this complexity. The average values for different classes of distances and angles are given in Table 4.

The molecular packing is illustrated in Fig. 6. The side chains and the benzoate groups project out from the five-membered ring to give a molecule extended in the (001) plane. The two side chains and the bromobenzoate groups A and B (Fig. 5) are in close contact whereas other molecules can approach the five-membered ring on both sides of group C. Similar parts of the molecules are in contact along the short c-axis, but this is not possible in other directions and carbon chains have to pack with benzene rings. However, the carbon chains group together as far as possible and alternate with regions of bromobenzoate groups. Some of the most important contacts are shown in Fig. 6 whereas contacts between one molecule and the molecule directly above or below in the c-direction are given in Table 5.

The benzene ring, the bromine atom and the carboxyl carbon are in all bromobenzoate groups planar within 0.08 Å. There is an indication that the carboxyl oxygen atoms may be slightly twisted out of the plane (up to 0.15 Å). Benzoate groups A and B are approximately parallel. Their planes make an angle of  $42^{\circ}$  with the (001) plane and their twofold axes are tilted 15° to the same plane. The third benzoate



Fig. 6. Molecular arrangement of the *p*-bromobenzoate derivative of the methyl ester of  $PGF_{2^{-1}}$  as seen along the *c*-axis with some of the more important contacts shown.

group shows very similar angles but has its diad axis roughly perpendicular to those of the other two groups when the three groups are in packing contact. Most benzene rings in this structure pack with their planes parallel but C-rings also make contact with roughly perpendicular planes at screw axes.



Fig. 7. Schematic representation of the hydrocarbon chain packing as seen along the chain axes. 0 and 1/2 are fractions of the repeat distance (2.55 Å) along a chain. Dotted line indicates the plane z=0.

The hydrocarbon chains are surprisingly regular considering their small length and their packing contacts with benzene rings. Atoms C<sub>21</sub> to C<sub>25</sub> are planar within 0.05 Å whereas C<sub>20</sub> is 0.16 Å off the plane. C<sub>39</sub> to C<sub>42</sub> lie within 0.002 Å from a plane with  $C_{38}$  at a distance of 0.12 Å. The axes of the two chains are roughly parallel to the (001) plane but are not mutually parallel as the position of the side chains in the molecule prevents them coming into closepacking contact until near their ends. The carbonchain arrangement is illustrated schematically in Fig. 7. The plane of chain D is inclined  $28^{\circ}$  to the (001) plane and that of chain  $E 76^{\circ}$  in the same sense. It is not to be expected that the chain packing should be exactly similar to those found in long-chain structures (cf. Abrahamsson et al., 1963) but the general arrangement of D-chains resembles very much that of the common orthorhombic type  $O_{\perp}$ .

The atoms of the ester group ( $C_{25}$ ,  $C_{26}$ ,  $O_5$ ,  $O_6$ ,  $C_{27}$ ) lie within 0.06 Å or less from a plane, which is twisted 23° from that of the adjoining hydrocarbon chain. A similar angle of twist (19°) is found in methyl stearate (Aleby & von Sydow, 1960).

Four of the atoms  $(C_8, C_9, C_{10}, C_{12})$  of the fivemembered ring lie in a plane within 0.07 Å whereas the fifth is at an appreciable distance (0.71 Å) as is usual for such saturated rings. The bonds to the different attached groups are all *trans* relative to the five membered ring (Fig. 5). The bonds  $C_{30}-O_8$  and  $C_{30}-C_{38}$  are rotated 45° and 54° respectively out of the plane of the double bonded group



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